

# Answer Key Ionic Bonds Compounds

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## **Ionic Bonds Answer Key arthurscience weebly com**

Ionic Bonds Answer Key Vocabulary chemical family electron affinity ion ionic bond metal nonmetal octet rule shell valence electron Prior Knowledge Questions Do these BEFORE using the Gizmo Note The purpose of these questions is to activate prior knowledge and get students thinking

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## **Answer Key Ionic Bonds and Compounds**

actually an ionic compound even though the formula does not contain a metal Although there is no metal present ammonium is still a cation electrons have been transferred and the attraction between ammonium and chloride is the same as if a metal cation were in the compound 24 Table salt is an ionic compound

## **from PO NO d2ct263enury6r cloudfront net**

ric Page 4 2 ANSWER KEY WRITING FORMULAS CRISS CROSS METHOD Name IONIC BONDING Name Ionic bonding occurs when a metal has lost one or more electrons 10 an on melo ll n on

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## **7 2 Ionic Bonds and Ionic Compounds gt CHEMISTRY YOU**

Formation of Ionic Compounds Ionic Bonds  $\hat{\sim}$  Sodium and chlorine atoms combine in a one to one ratio and both ions have stable octets When sodium and chlorine react to form a compound the sodium atom transfers its one valence electron to the chlorine atom

## **Ionic Bonding Practice answer key BetterLesson**

Guided Practice I ask students to write the salts for problems 2 4 from the Ionic Bonding Practice problems I then show the class the answers using the Ionic Bonding Practice answer key This balancing charges in salts video shows students explaining how to do this skill

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## **Chapter 8 Covalent Bonding Union County Vocational**

Single Covalent Bonds When only one pair of electrons is shared such as in a hydrogen molecule it is a single covalent bond The shared electron pair is often referred to as the bonding pair For a hydrogen molecule shown in Figure 8 4 each covalently bonded atom equally attracts the pair of shared electrons

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